

Published on Web 01/17/2003

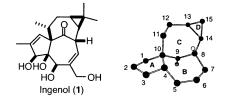
Total Synthesis of Ingenol

Keiji Tanino,*,[†] Kei Onuki,[†] Kohei Asano,[†] Masaaki Miyashita,[†] Tsuyoshi Nakamura,[§] Yoshinori Takahashi,[§] and Isao Kuwajima*,[‡]

Division of Chemistry, Graduate School of Science, Hokkaido University, Sapporo 060-0810, Japan, Department of Chemistry, Tokyo Institute of Technology, Meguro, Tokyo 152-8551, Japan, and Laboratory for Natural Products Chemistry, Kitasato University, and CREST, Japan Science and Technology Corporation (JST), S-105 1-15-1, Kitasato, Sagamihara 228-8555, Japan

Received November 5, 2002; E-mail: ktanino@sci.hokudai.ac.jp

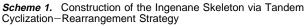
Ingenol (1),¹ isolated from the genus *Euphorbia*, has been of great interest as a synthetic target² because of its unusual structure involving an "inside—outside" bridged BC ring coupled with a broad spectrum of biological activities.³ The difficulties in constructing the highly strained ingenane skeleton required special approaches, and five strategically distinct methods^{2,4} have been disclosed to date. The first total synthesis of ingenol has recently been achieved by Winkler⁵ on the basis of an intramolecular de Mayo reaction that had been originally reported by the same authors⁶ as the first entry for constructing the trans-bicyclo[4.4.1]-undecane system.

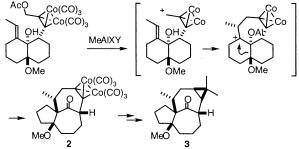


In 1997, we reported an efficient method for ingenane synthesis by employing a "tandem cyclization—rearrangement strategy"⁷ as shown in Scheme 1. Although the product **2** was further converted into compound **3** having the complete CD ring system of ingenol,⁸ the methoxy group at the C(4) position was not sufficient to serve as an anchor for installation of the hydroxyl groups and the olefin moieties of the AB ring.

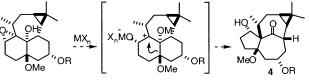
The present work describes the total synthesis of ingenol on the basis of a new strategy, namely a rearrangement reaction of an epoxy alcohol giving rise to the key intermediate **4** which possesses two oxygen functional groups on both the A and B rings (Scheme 2).

The synthetic route for the key intermediate is depicted in Scheme 3. An allyl ether derived from commercially available alcohol **5** was subjected to a Claisen rearrangement reaction promoted by dichloroacetic acid. The resulting ketone **6** reacted with NBS in water–DMSO⁹ to yield bromohydrin derivatives **7** and **8** as an almost 1:1 mixture, which was recrystallized from ethyl acetate to give pure **8** as crystals. It is noteworthy that ketone **6** can be easily recovered from the mother liquid containing **7** by treating with zinc dust. Silylation of **8** by using TESCl and imidazole effected selective protection of the secondary hydroxyl group via opening of the lactol ring. The aldol reaction of ketone **9** with acetaldehyde followed by dehydration afforded (*E*)-enone **10** stereoselectively. In the presence of LiBr, enone **10** reacted with lithium enolate of *tert*-butyl acetate to afford adduct **11** as a single









isomer. In the stereoselective addition reaction, the conformation of the α -methoxyketone would be restricted by forming a fivemembered chelate ring, and the nucleophile attacks the carbonyl group from the opposite side of the side chain (Figure 1, A).

Treatment of hydroxyester 11 with trimethylaluminum followed by LDA effected an intramolecular alkylation reaction to give transdecaline derivative 12 as a single isomer. Since a reaction in the absence of trimethylaluminum failed to give 12, formation of a six-membered cyclic aluminum enolate intermediate would be essential for the stereoselective cyclization reaction (Figure 1, B). Silvlation of the secondary hydroxyl group followed by elongation of the side chain led to a propargyl alcohol which was converted into dicobalt acetylene complex 15 in good yield. Under the influence of methylaluminum bis(2,6-dimethyl-4-nitrophenoxide), cobalt complex 15 underwent a cyclization reaction to afford allyl alcohol 16 containing the C(11) α -methyl group. The dicobalt acetylene complex moiety of 16 was used for stereoselective construction of the D ring through Birch reduction, dibromocyclopropanation, and methylation. Transformation of the tetracyclic carbon framework into an ingenane skeleton was achieved via stereoselective epoxidation of allyl alcohol 17 followed by treatment with trimethylaluminum.¹⁰

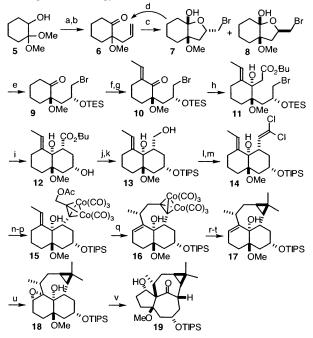
The key intermediate **19** having a complete ABCD ring system of ingenol in hand, the stage was set for stereoselective functionalization of the AB ring moiety (Scheme 4). The condensation reaction of ketone **20** with *tert*-butoxy-bis(dimethylamino)methane¹¹ followed by reduction with DIBAL¹² afforded enone **21** in an excellent yield. The C(1)–C(2) double bond was introduced by dehydration of a secondary alcohol which was obtained through

[†] Hokkaido University.

[§] Tokyo Institute of Technology.

[‡] Kitasato University.

Scheme 3. Construction of the Ingenane Skeleton via Rearrangement Strategy^a



^{*a*} Reagents and conditions: (a) NaH, BrCH₂CH=CH₂, benzene; (b) CHCl₂CO₂H, DMF, reflux (79% in 2 steps); (c) NBS, H₂O, DMSO (8: 32% after recrystallization); (d) Zn dust, NH₄Cl, ether, H₂O (52% of **6** was recovered); (e) TESCl, imidazole, DMF, rt (99%); (f) LHMDS, ZnBr₂, CH₃CHO, THF (71%); (g) (CF₃CO)₂O, pyridine, DBU, benzene (95%); (h) 'BuOAc, LDA, LiBr, ether (91%); (i) Me₃Al, LDA, THF, then HCl (66%); (j) TIPSCl, DMAP, DMF (quant); (k) LiAlH₄, THF (84%); (l) SO₃-pyridine, DMSO, Et₃N (97%); (m) (EtO)₂P(=O)CCl₃, BuLi, THF (85%); (n) BuLi, THF, then (CH₂O)_n (91%); (o) Ac₂O, Et₃N, DMAP, CH₂Cl₂ (99%); (p) Co₂(CO)₈, CH₂Cl₂ (95%); (q) methylaluminum bis(2,6-dimethyl-4-nitrophenoxide CH₂Cl₂, r) Li, liq. NH₃ (67% in 2 steps); (s) CHBr₃, NaOH, BnEt₃NCl, CH₂Cl₂, H₂O (71%); (t) Me₃CuLi₂, ether, then MeI (95%); (u) TBHP, Ti(O'Pr)₄, MS 4A, CH₂Cl₂; (v) Me₃Al, CH₂Cl₂ (76% in 2 steps).

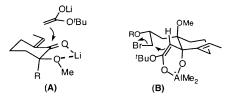
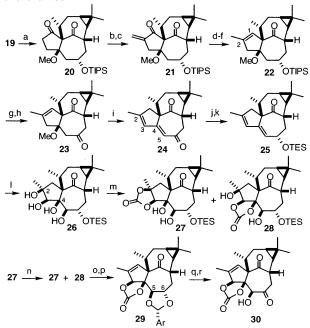


Figure 1. Suggested intermediates of the stereoselective C–C bond-forming reactions.

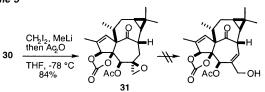
successive treatment of enone 21 with sodium borohydride and DIBAL. Next, silyl ether 22 was converted into the corresponding ketone with a view to introducing the C(4)-C(5) double bond by a β -elimination reaction. Under the influence of a guanidine base, however, ketone 23 yielded conjugated diene 24 via isomerization of the C(1)-C(2) double bond. The result led us to examine construction of the polyol moiety by osmium tetroxide oxidation of both C(2)-C(3) and C(4)-C(5) double bonds in one stage. Since the oxidation reaction of enone 24 resulted in cleavage of the B ring via a retro aldol reaction, the C(6) carbonyl group was masked by stereoselective reduction and silvlation. Treatment of diene 25 with an excess amount of osmium tetroxide followed by sodium hydrogen sulfite gave polyol 26, containing all of the stereogenic centers of ingenol, as a single isomer. To accomplish functionalization of the A ring, it was then required to protect the C(3), C(4), and C(5) hydroxyl groups selectively. The reaction of 26 with 1,1'carbonyldiimidazole yielded a ca. 4:1 mixture of cyclic carbonates 27 and 28, which can be easily separated by silica gel column

Scheme 4. Stereoselective Introduction of the Oxygen Functionalities^a



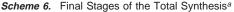
^{*a*} Reagents and conditions: (a) DMSO, (COCl)₂, Et₃N, CH₂Cl₂ (93%);. (b) 'BuO(Me₂N)₂CH, DMF, 100 °C; (c) DIBAL, CH₂Cl₂, MeI, THF (98% in 2 steps); (d) NaBH₄, EtOH (95%); (e) DIBAL, CH₂Cl₂; (f) (CF₃SO₂)₂O, 2,6-lutidine, DBU, CH₂Cl₂ (83% in 2 steps); (g) TBAF, THF (100%); (h) PDC, CH₂Cl₂ (97%); (i) 1,3,4,6,7,8-hexahydro-1-methyl-2H-pyrimido[1,2-a]pyrimidine, DMF, 120 °C (86%); (j) NaBH₄, CeCl₃, MeOH-H₂O; (k) TESCl, imidazole, DMF (99% in 2 steps); (l) OsO₄, pyridine, ether, then NaHSO₃, H₂O (59%); (m) 1,1'-carbonyldiimidazole, toluene (**27**: 76%, **28**: 18%); (n) 4-(dimethyl-amino)pyridine, toluene, 100 °C; (**27**: 72%, **28**: 27%) (o) *p*-MeOC₆H₄CH(OMe)₂, CSA, DMF (93%); (p) SOCl₂, pyridine; (q) AcOH-H₂O (96% in 2 steps); (r) Me₂S, NCS, toluene (75%).

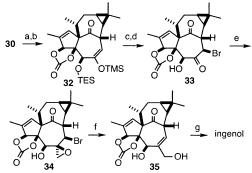




chromatography. Although the desired compound **28** was the minor product, **27** was found to undergo isomerization to give a ca. 3:1 mixture of **27** and **28** simply by heating with 4-(dimethylamino)-pyridine. Removal of the triethylsilyl group and protection of the C(5) and C(6) hydroxyl groups as a cyclic acetal was achieved in one pot, and the resulting *tert*-alcohol was subjected to a dehydration reaction to afford olefin **29** as a single regioisomer. Since attempts for distinction between the C(5) and C(6) hydroxyl groups via regioselective cleavage of the cyclic acetal moiety were fruitless, **29** was hydrolyzed to the corresponding C(5)–C(6) diol. Fortunately, the diol was selectively converted into α -ketol **30** by adopting the Corey–Kim protocol¹³ through oxidation of the less hindered C(6) hydroxyl group.

Construction of the allylic alcohol moiety by using the C(6) keto group remained as the final step in the total synthesis. We initially chose epoxide **31**, which was prepared from ketol **30** by successive treatment with diiodomethane, methyllithium,¹⁴ and acetic anhydride, as a precursor of an allylic alcohol (Scheme 5). However, the conventional methods using an aluminum amide,¹⁵ aluminum isopropoxide,¹⁶ or TMSOTf-DBU¹⁷ failed to effect the desired





^a Reagents and conditions: (a) TESOTf, 2,6-lutidine, CH₂Cl₂; (b) TMSCl, Et₃N, LDA, THF; (c) NBS, CH₂Cl₂; (d) HF, CH₃CN; (e) CH₂I₂, MeLi, THF (39% in 5 steps); (f) Zn, NH₄Cl, THF-H₂O (91%); (g) KOH, MeOH (89%).

isomerization reaction. These results led us to examine an alternative method for ring-opening, namely, reductive cleavage of a halo epoxide.18

With a view to introducing a bromine at the C(7) position, ketol 30 was converted into enol silvl ether 32. In these reactions, the α -siloxyketone intermediate showed unusual lability and gave a complex mixture even by silica gel column chromatography. Although the α -siloxyketone also underwent complete decomposition in the presence of LDA, it was found that addition of LDA to a mixture of the crude α -siloxyketone and TMSCl effects the desired transformation.

The reaction of 32 with NBS19 followed by desilylation afforded α -bromoketone 33 that was converted into epoxide 34 by adopting a method similar to that described above. Treatment of 34 with zinc dust and an aqueous NH₄Cl solution induced reductive cleavage of the bromo epoxide moiety, and the resulting allylic alcohol 35 was subjected to hydrolysis to afford ingenol, which gave spectral data in full agreement with those of the natural product.

In conclusion, total synthesis of ingenol was accomplished on the basis of novel key reactions, namely an intramolecular cyclization reaction of acetylene dicobalt complex 15, a rearrangement reaction of epoxy alcohol 18 for constructing the ingenane skeleton, and a stereoselective double dihydroxylation reaction of diene 25. The 45-step transformation from commercially available compound 5 afforded ingenol in ca. 0.1% overall yield. Further studies on asymmetric total synthesis of ingenol are currently under way in our laboratory.

Acknowledgment. This research was supported in part by the Ministry of Education, Culture, Sports, Science and Technology.

We also acknowledge financial support from The Sumitomo Foundation, Novartis Foundation (Japan) for the Promotion of Science, and The Akiyama Foundation. We thank Professor G. Appendino (Dipartimento di Scienze Chimiche, Alimentari, Farmaceutiche e Farmacologiche) for providing natural ingenol.

Supporting Information Available: Experimental details and spectroscopic data (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (a) Hecker, E. Cancer Res. 1968, 28, 2338. (b) Zechmeister, K.; Brandl, ; Hoffe, W.; Hecker, E.; Opferkuch, H. J.; Adolf, W. Tetrahedron Lett. 1970, 4075. See also; (c) Uemura, D.; Hirata, Y. Tetrahedron Lett. 1971, 3673
- (2) For recent reviews on synthetic studies on ingenol, see: (a) Rigby, J. H. Stud. Nat. Prod. Chem. 1993, 12, 233. (b) Kim, S.; Winkler, J. D. Chem. Soc. Rev. 1997, 26, 387.
- (3) For examples, (a) Kupchan, S. M.; Uchida, I.; Branfman, A. R.; Dailey, R. G.; Fei, B. Y. *Science* **1976**, *191*, 571. (b) Hecker, E. *Pure Appl. Chem.* **1977**, *49*, 1423. (c) Hasler, C. M.; Acs, G.; Blumberg, M. *Cancer Res.* **1992**, *52*, 202. (d) Fujiwara, M.; Ijichi, K.; Tokuhisa, K.; Katsuura, K.; Shigeta, S.; Konno, K.; Wang, G.-Y.-S.; Uemura, D.; Yokota, T.; Baba, M. Antimicrob. Agents Chemother. 1996, 40, 271.
- (4) For recent reports that are not covered in ref 2, see: (a) Winkler, J. D.; (4) For recent reports that are not covered in ref 2, see: (a) Winkler, J. D.; Kim, S.; Harrison, S.; Lewin, N. E.; Blumberg, P. M. J. Am. Chem. Soc. 1999, 121, 296. (b) Kigoshi, H.; Suzuki, Y.; Aoki, K.; Uemura, D. Tetrahedron Lett. 2000, 41, 3927. (c) Tang, H.; Yusuff, N.; Wood, J. L. Org. Lett. 2001, 3, 1563. (d) Rigby, J. H.; Bazin, B.; Meyer, J. H.; Mohammadi, F. Org. Lett. 2002, 4, 799.
 (5) Winkler, J. D.; Rouse, M. B.; Greaney, M. F.; Harrison, S. J.; Jeon, Y. T. Law, Chem. Chem. Cover. 2002.
- J. Am. Chem. Soc. 2002, 124, 9726.
- (6) Winkler, J. D.; Henegar, K. E. J. Am. Chem. Soc. 1987, 109, 2850. (7) Nakamura, T.; Matsui, T.; Tanino, K.; Kuwajima, I. J. Org. Chem. 1997,
- 62.3032
- (8) Tanino, K.; Nakamura, T.; Matsui, T.; Kuwajima, I. Tennen Yuki Kagobutsu Toronkai Koen Yoshishu 1997, 39, 1.
- (9)Dalton, D. R.; Dutta, V. P.; Jones, D. C. J. Am. Chem. Soc. 1968, 90, 5498
- (10) For examples of ring-enlargement reactions via rearrangement of an epoxy alcohol, see: (a) Maruoka, K.; Hasegawa, M.; Yamamoto, H.; Suzuki, K.; Shimazaki, M.; Tsuchihashi, G. J. Am. Chem. Soc. 1986, 108, 3827. (b) Tu, Y. Q.; Sun, L. D.; Wang, P. Z. J. Org. Chem. 1999, 64, 629.
- (11) Bredereck, H.; Simchen, G.; Rebsdat, S.; Kantlehner, W.; Horn, P.; Wahl, R.; Hoffman, H.; Grieshaber, P. Ber. 1968, 101, 41. See also: Trast, B. M.; Preckel, M. J. Am. Chem. Soc. 1973, 95, 7862.
- (12) Murai, A.; Tanimoto, N.; Sakamoto, N.; Masamune, T. J. Am. Chem. Soc. 1988, 110, 1985.
- (13) Corey, E. J.; Kim, S. U. J. Am. Chem. Soc. 1972, 94, 7586.
- (14) (a) Tarhouni, R.; Kirschleger, B.; Rambaud, M.; Villieras, J. Tetrahedron Lett. 1984, 25, 835. (b) Sadhu, K. M.; Matteson, D. S. Tetrahedron Lett. 1986, 27, 795. Use of diiodomethane instead of dibromomethane or chloroiodomethane effected rapid formation of the epoxide at a low temperature
- (15) Yasuda, A.; Tanaka, S.; Oshima, K.; Yamamoto, H.; Nozaki, H. J. Am. (15) Tustua, A., Tana, S., S., S., Chem. Soc. 1974, 96, 6513.
 (16) Eschinasi, E. H. J. Org. Chem. 1970, 35, 1598.
- (17) Murata, S.; Suzuki, M.; Noyori, R. J. Am. Chem. Soc. 1979, 101, 2738.
- (18) For a similar transformation, see: Concellón, J. M.; Llavona, L.; Bernad, P. L. *Tetrahedron* **1995**, *51*, 5573.
- (19) Reuss, R. H.; Hassner, A. J. Org. Chem. 1974, 39, 1785.

IA029226N